First Year Pharm D Degree Examinations - August 2013

Pharmaceutical Inorganic Chemistry

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1.	Explain the principle behind non aqueous titrations, taking the example of titration of halogen salts of bases.		
2.	Discuss the theory behind complexometric titrations. Mention the details of the		
	titrants and indicators used in complexometric titrations.		(5+3+2)
3.	Explain the principle	e behind the limit tests for sulphates and the limit test f	or iron (4+6)
Short notes: (8x5=40)			(8x5=40)
4.	Discuss the differer	it types of errors.	

- 5. Explain the following terms and mention one example each:
 - Saline cathartics **Dentifrices** • •
- 6. Explain how hydrogen peroxide is assayed as per the method mentioned in the IP
- 7. Explain how a sample of boric acid is assayed as per the IP.
- 8. Explain the rationale behind combining dried aluminium hydroxide gels with magnesium hydroxide gels in antacid formulations.
- 9. Discuss the physiological role of zinc and manganese in human health.
- 10. Mention the method of assay of oxygen and the storage conditions.
- 11. Name five substances used as pharmaceutical aid for the specific purpose as:
 - Anti oxidants Absorbents and adsorbents • •

Time: 3 Hours

Total Marks: 70